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(57) Abstract:

A new series of metal complexes (Zn(II), Cd(II), Ni(II), Cu(II), Fe(III), Co(II), Mn(II) Hg(II), and Ag(I)) have been synthesized using Schiff base derived from 3,4-(methylenedioxy)aniline and 5-chloro salicylaldehyde. Their structures were confirmed by elemental analysis, spectroscopic and molar conductivity studies. The complexes are of the type 1:1(metal:ligand) ratio. In all cases, the Schiff base is bonded in a bidentate fashion through azomethine nitrogen and phenolic oxygen. Coordination sphere is completed by one Cl or NO3 or CH3COO group and one water molecule in a tetrahedral arrangement. Fe (III) complex alone shows octahedral arrangement with two Cl groups and two water molecules. The proposed structures for these complexes are further supported by thermal and mass spectral analysis. The biological activities of all the synthesized complexes were screened against five bacterial (E.coli, S.aureus, E.faecalis, P.fluorescens, and Klebsiella sp.) and three fungal (Candidaalbicans, Fusarium.sp and Trichosporon.sp) strains. All the complexes exhibit increased activity (bacterial, fungal and antioxidant) compared to the Schiff base ligand.

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